The Punjab School-DRC Class 9th

Chemistry- Vacation Engaging Task

Chapter 1

- 1. Write definitions of following and also write their scope and examples.
 - a) Definition of chemistry.
 - **b)** Physical chemistry
 - c) Organic chemistry
 - d) Inorganic chemistry
 - e) Bio-chemistry
 - f) Industrial chemistry
 - g) Nuclear chemistry
 - h) Industrial chemistry
 - i) Nuclear chemistry
 - j) Environmental chemistry
 - k) Analytical chemistry
- 2 .Difference b/w substance & element.
- 3. What are physical and chemical properties? Also give examples.
- 4. Define elements and write their classification.
- 5. What is valency? Explain with examples.
- 6. Name the elements that show variable valency and give examples.
- 7. What is compound? Also give examples.
- 8. Differentiate b/w compound and mixture.
- 9. What are homogeneous mixture and heterogeneous mixture? Also give examples.

Chapter 5

- **1.** What is matter and write the names of three states of matter.
- 2. What are solids, liquids, and gases?
- 3. Differentiate diffusion and effusion?
- **4.** What is pressure and write its units?
- 5. Define standard atmospheric pressure?
- 6. Why gases are compressible?
- 7. Define Boyle's law.
- 8. Define Charle's law.
- **9.** What is systolic and diastolic pressure?
- 10. What is evaporation?
- **11.** Why evaporation cause cooling?
- 12. Define vapour pressure?
- 13. Define boiling point and freezing point.
- **14.** Which principle is used in "Pressure Cooker"?
- 15. Define amorphous solids with example.
- **16.** Define crystalline solids with example.
- **17.** What is allotropy? What are reasons of allotropy?
- **18.** Define transition temperature with examples.
- **19.** Which form of Sulphur exists at room temperature?
- **20.** Why is white tin available at room temperature?
- **21.** Why amorphous solids do not have sharp melting points while crystalline solids do have?